

(19)



Europäisches Patentamt
European Patent Office
Office européen des brevets



(11)

EP 1 081 209 A1

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:
07.03.2001 Bulletin 2001/10

(51) Int Cl.7: **C10L 1/22, C08F 8/32,
C08G 14/06, C10L 10/00,
C10L 10/04, C10L 1/14**

(21) Application number: **00307444.0**

(22) Date of filing: **30.08.2000**

(84) Designated Contracting States:
**AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE**
Designated Extension States:
AL LT LV MK RO SI

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(30) Priority: **31.08.1999 US 386690**

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(54) **Mannich dispersants with enhanced lubricity**

(57) A product obtainable by reacting:

- (i) a compound containing at least one primary or secondary amine group and at least one hydroxyl group;
- (ii) a polyamine containing at least two amine groups, which is different from (i), wherein at least

one of the amine groups is a primary or secondary amine;

- (iii) an alkyl-substituted hydroxyaromatic compound; and
- (iv) at least one aldehyde,

is useful in a hydrocarbon fuel as a detergent and also to enhance the lubricity of the fuel.

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Description

TECHNICAL FIELD

5 [0001] The present invention relates to fuel compositions comprising a dispersant additive for hydrocarbon fuels, such as gasoline and diesel fuel, wherein said dispersant additive is the reaction product of i) a compound containing at least one primary or secondary amine group and at least one hydroxyl group; ii) a polyamine containing at least two amine groups wherein at least one of the amine groups is a primary or secondary amine, and wherein component ii) is different from i); iii) an alkyl-substituted hydroxyaromatic compound; and iv) an aldehyde.

BACKGROUND OF THE INVENTION

15 [0002] It has long been desired to maximize fuel economy and power in diesel engines while enhancing acceleration and preventing knocking and hesitation. Diesel fuel must ignite spontaneously and quickly (within 1 to 2 milliseconds) without a spark. The time lag between the initiation of injection and the initiation of combustion is called ignition delay. In high-speed diesel engines, a fuel with a long ignition delay tends to produce rough operation and knocking.

[0003] Two major factors affect ignition delay: the mechanical conditions in the engine and the chemistry of the fuel. The mechanical component is influenced by such factors as compression ratio, motion of the air charge during ignition and ability of the fuel injector to atomize fuel. The chemical component of ignition delay is influenced by such factors as the fuel's autoignition temperature, specific heat, density, viscosity, and other properties. The ability of a diesel fuel to ignite quickly after injection into a cylinder is known as its cetane number.

[0004] To minimize ignition delay in a diesel engine, it is desirable to enhance the mechanical component by maintaining the fuel injector's ability to precisely atomize fuel by keeping the injectors clean. However, this must be done in such a way that does not negatively affect the chemical component.

25 [0005] It would be beneficial to provide dispersant compositions that are highly effective in minimizing injector deposits in diesel engines, as well as intake valve deposits in gasoline engines, and that provide such an advantage without harm to the engine and without deterioration in engine performance.

[0006] Accordingly, one of the objects of the present invention is to provide a diesel fuel that provides effective detergency without attendant deterioration in engine performance.

30 [0007] Problems associated with fuel lubricity arose in the mid-1960's when a number of aviation fuel pump failures occurred. After considerable research, it was realized that advances in the refining of aviation turbine fuel had resulted in the almost complete removal of the naturally occurring lubricating components from the fuel. The removal of these natural lubricants resulted in the seizure of fuel pump parts. By the mid-1980's, it seemed likely that a similar problem was imminent in diesel fuel pumps. Fuel injection pump pressures had been steadily increasing while there was also a growing concern to reduce the sulfur content of the diesel fuel. The desire to reduce the sulfur content of the diesel fuel, in an effort to reduce pollution, required the use of more rigorous fuel refining processes. It was determined that as refining processes became more stringent, the naturally occurring sulfur, nitrogen and oxygen containing compounds and polyaromatics which contribute to diesel fuel's inherent lubricity were reduced or eliminated. In response to these developments, a number of effective lubricity additives were developed for diesel fuels. These additives are now widely used to enhance the lubricity of highly refined, low sulfur diesel fuels.

40 [0008] In certain types of in-line diesel injection pumps, engine oil contacts diesel fuel. Engine oil may also come into contact with the diesel fuel through direct addition of used engine oil to the fuel. Certain types of lubricity additives used in low sulfur diesel fuel have been found to contribute to fuel filter blockage and to pump plunger sticking. Lubricity additives having poor compatibility with engine oil have been shown to cause these problems. Compatibility is defined as the tendency for the diesel fuel containing the lubricity additive not to form fuel insoluble deposits, gels or heavy sticky residues when in contact with engine oil. These deposits, gels or residues have been shown to cause fuel filter blockage and injection pump sticking. The additives of the present invention are compatible with engine oil.

45 [0009] Gasoline fuels are also becoming subject to compositional constraints, including restrictions on sulfur content, in an effort to reduce pollutants. The principal concern is the effect of sulfur on exhaust catalyst life and performance. The lubricity requirements of gasoline are somewhat lower than for diesel fuel since the majority of gasoline fuel injection systems inject fuel upstream of the inlet valves and thus operate at much lower pressures than diesel fuel pumps. However, as automobile manufacturers desire to have electrically powered fuel pumps within the fuel tanks, failure of the pumps can be expensive to repair. These problems are also likely to increase as injection systems become more sophisticated and the gasoline fuels become more highly refined.

55 [0010] Additional pump wear concerns have arisen with the introduction of vehicles having direct injection gasoline (DIG) engines since the fuel pumps for these vehicles operate at significantly higher pressures than traditional gasoline fuel pumps.

[0011] Another area subject to pump wear and failure is the use of submerged fuel pumps in gasoline or diesel fuel

storage tanks. It is important to reduce the wear of these submerged pumps due to the difficulty of accessing these pumps for repair and maintenance.

[0012] Many commercially available gasoline fuels contain gasoline detergents such as polyisobutylene amine and polyether amine. These compounds are known to have a minor effect on the wear properties of the fuel. A growing number of commercially available gasoline fuels contain oxygenates, such as methyltertiarybutylether (MTBE). These oxygenates are known to increase rates of wear of fuel pump components as they have very high friction coefficients. In light of the desire for more highly refined fuels, lower sulfur content and oxygenation of the fuels, there is presently a need for lubricity improvers for hydrocarbon fuels in order to obtain acceptable fuel pump life. The present invention addresses these problems by adding the novel Mannich reaction products to the fuel.

[0013] While the prior art is replete with numerous treatments for fuels, it does not disclose the addition of the present additives to hydrocarbon fuels or teach their use for providing enhanced detergency and lubricity to said fuels.

SUMMARY OF THE INVENTION

[0014] The present invention relates to the treatment of a hydrocarbon fuel to substantially reduce the wear occasioned upon fuel pumps used to pump said hydrocarbon fuels and to provide effective detergency to the fuels without attendant deterioration in engine performance. The present invention also relates to the discovery that the addition to a fuel of the reaction products of the present invention will improve detergency and lubricity as compared to a similar fuel that has not been treated with said reaction products.

[0015] Thus, there is disclosed a fuel composition comprising a major amount of a hydrocarbon fuel and a minor amount of a fuel-soluble dispersant obtained by reacting i) a compound containing at least one primary or secondary amine group and at least one hydroxyl group; ii) a polyamine containing at least two amine groups wherein at least one of the amine groups is a primary or secondary amine, and wherein component ii) is different from i); iii) an alkyl-substituted hydroxyaromatic compound; and iv) an aldehyde. The dispersant additives are preferably present in the fuel in an amount within the range of from about 1 to about 1000 parts by weight of additive per million parts by weight of fuel (ppm w/w). More preferably, the dispersants are present in the fuel in an amount within the range from about 10 to about 500 ppm w/w, most preferably, from about 30 to about 300 ppm w/w.

[0016] There is also disclosed a method for reducing the wear of fuel pumps through which a hydrocarbon fuel is pumped, comprising adding a fuel-soluble additive to said fuel wherein the fuel-soluble additive comprises the above-described dispersant and wherein the dispersant additive is added to the fuel in an amount effective to improve the detergency and lubricity of the fuel, typically, the dispersant additive is present in the fuel composition in an amount of at least 1 ppm.

[0017] In view of the problems discussed above, a general aspect of the present invention is to provide a fuel additive that gives improved detergency to the fuel and protects the fuel pump from excessive wear and breakdown. A further aspect of the invention is to provide a fuel additive suitable for addition to a fuel that does not damage the fuel system and does not cause an increase in undesirable combustion products.

DETAILED DESCRIPTION OF THE INVENTION

[0018] The dispersant additives of the present invention may be categorized as Mannich reaction products and are fuel-soluble reaction products obtained by the reaction of i) a compound containing at least one primary or secondary amine group and at least one hydroxyl group; ii) a polyamine containing at least two amine groups wherein at least one of the amine groups is a primary or secondary amine, and wherein component ii) is different from i); iii) an alkyl-substituted hydroxyaromatic compound; and iv) an aldehyde.

[0019] Amines suitable for use as component i) contain at least one primary or secondary amine group and at least one hydroxyl group. The preferred amine reactants for use as component i) in the present invention are amino-alcohols; alkoxyated amines; and mixtures thereof. Examples of suitable amino-alcohols include ethanolamine and diethanolamine; representative alkoxyated amines include ethoxyated and propoxyated amines and polyamines. An example of these amines includes, for example, 2-(2-aminoethylamino) ethanol. A particularly preferred amine for use as component i) is diethanolamine.

[0020] Polyamines suitable for use as component ii) contain at least two amine groups wherein at least one of the amine groups is a primary or secondary amine, and wherein component ii) is different from i).

[0021] Representative amine reactants useful as component ii) include, but are not limited to, alkylene polyamines having at least one suitably reactive primary or secondary amino group in the molecule. Other substituents may be present in the polyamine. In a preferred embodiment, the alkylene polyamine is a polyethylene polyamine. Suitable alkylene polyamine reactants include ethylenediamine, diethylenetriamine, triethylenetetramine, tetraethylenepentamine, pentaethylenhexamine, hexaethylenheptamine, heptaethylene octamine, octaethylenenonamine, nonaethylenedecamine, decaethylenundecamine and mixtures of such amines having nitrogen contents correspond-

ing to alkylene polyamines of the formula $H_2N-(A-NH-)_nH$, where A is divalent ethylene or propylene and n is an integer of from 1 to 10. The alkylene polyamines may be obtained by the reaction of ammonia and dihalo alkanes, such as dichloro alkanes. Thus, the alkylene polyamines obtained from the reaction of 2 to 11 moles of ammonia with 1 to 10 moles of dichloro alkanes having 2 to 6 carbon atoms and the chlorines on different carbon atoms are suitable alkylene polyamine reactants.

[0022] In another preferred embodiment of the present invention, the amine useful as component ii) is a polyamine having at least one primary or secondary amino group and at least one tertiary amino group in the molecule. Examples of suitable polyamines include N,N,N',N'-tetraalkyldialkylenetriamines (two terminal tertiary amino groups and one central secondary amino group), N,N,N',N'-tetraalkyltrialkylenetetramines (one terminal tertiary amino group, two internal tertiary amino groups and one terminal primary amino group), N,N,N',N',N''-pentaalkyltrialkylenetetramines (one terminal tertiary amino group, two internal tertiary amino groups and one terminal secondary amino group), tris(di-alkylaminoalkyl)aminoalkylmethanes (three terminal tertiary amino groups and one terminal primary amino group), and like compounds, wherein the alkyl groups are the same or different and typically contain no more than about 12 carbon atoms each, and which preferably contain from 1 to 4 carbon atoms each. Most preferably these alkyl groups are methyl and/or ethyl groups. Preferred polyamine reactants include N,N-dimethyl-1,3-propanediamine and N-methyl piperazine.

[0023] The alkyl-substituted hydroxyaromatic compounds and aldehydes used in making the Mannich detergents of the present invention may be any such compounds known and applied in the art, in accordance with the foregoing limitations.

[0024] Representative alkyl-substituted hydroxyaromatic compounds that may be used in forming the present Mannich products are polypropylphenol (formed by alkylating phenol with polypropylene), polybutylphenols (formed by alkylating phenol with polybutenes and/or polyisobutylene), and polybutyl-co-polypropylphenols (formed by alkylating phenol with a copolymer of butylene and/or butylene and propylene). Other similar long-chain alkylphenols may also be used. Examples include phenols alkylated with copolymers of butylene and/or isobutylene and/or propylene, and one or more mono-olefinic comonomers copolymerizable therewith (e.g., ethylene, 1-pentene, 1-hexene, 1-octene, 1-decene, etc.) where the copolymer molecule contains at least 50% by weight, of butylene and/or isobutylene and/or propylene units. The comonomers polymerized with propylene or such butenes may be aliphatic and can also contain non-aliphatic groups, e.g., styrene, o-methylstyrene, p-methylstyrene, divinyl benzene and the like. Thus in any case the resulting polymers and copolymers used in forming the alkyl-substituted hydroxyaromatic compounds are substantially aliphatic hydrocarbon polymers.

[0025] Polybutylphenol (formed by alkylating phenol with polybutylene) is preferred. Unless otherwise specified herein, the term "polybutylene" is used in a generic sense to include polymers made from "pure" or "substantially pure" 1-butene or isobutene, and polymers made from mixtures of two or all three of 1-butene, 2-butene and isobutene. Commercial grades of such polymers may also contain insignificant amounts of other olefins. So-called high reactivity polybutylenes having relatively high proportions of polymer molecules having a terminal vinylidene group, formed by methods such as described, for example, in U.S. Pat. No. 4,152,499 and W. German Offenlegungsschrift 29 04 314, are also suitable for use in forming the long chain alkylated phenol reactant.

[0026] The alkylation of the hydroxyaromatic compound is typically performed in the presence of an alkylating catalyst at a temperature in the range of about 50 to about 200 °C. Acidic catalysts are generally used to promote Friedel-Crafts alkylation. Typical catalysts used in commercial production include sulphuric acid, BF_3 , aluminum phenoxide, methanesulphonic acid, cationic exchange resin, acidic clays and modified zeolites.

[0027] The long chain alkyl substituents on the benzene ring of the phenolic compound are derived from polyolefin having a number average molecular weight (M_n) of from about 500 to about 3000 (preferably from about 500 to about 2100) as determined by gel permeation chromatography (GPC). It is also preferred that the polyolefin used have a polydispersity (weight average molecular weight/number average molecular weight) in the range of about 1 to about 4 (preferably from about 1 to about 2) as determined by GPC.

[0028] The chromatographic conditions for the GPC method referred to throughout the specification are as follows: 20 μ L of sample having a concentration of approximately 5 mg/mL (polymer/unstabilized tetrahydrofuran solvent) is injected into 1000A, 500A and 100A columns at a flow rate of 1.0 mL/min. The run time is 40 minutes. A Differential Refractive Index detector is used and calibration is relative to polyisobutene standards having a molecular weight range of 284 to 4080 Daltons.

[0029] The Mannich dispersant may be, and preferably is, made from a long chain alkylphenol. However, other phenolic compounds may be used including high molecular weight alkyl-substituted derivatives of resorcinol, hydroquinone, catechol, hydroxydiphenyl, benzylphenol, phenethylphenol, naphthol, tolynaphthol, among others. Preferred for the preparation of the Mannich condensation products are the polyalkylphenol reactants, e.g., polypropylphenol and polybutylphenol whose alkyl group has a number average molecular weight of about 500 to about 2100, while the most preferred alkyl group is a polybutyl group derived from polybutylene having a number average molecular weight in the range of about 800 to about 1300.

[0030] The preferred configuration of the alkyl-substituted hydroxyaromatic compound is that of a para-substituted mono-alkylphenol. However, any alkylphenol readily reactive in the Mannich condensation reaction may be employed. The long chain alkyl substituents may contain some residual unsaturation, but in general, are substantially saturated alkyl groups.

[0031] Representative aldehydes for use in the preparation of the detergent additives include the aliphatic aldehydes such as formaldehyde, acetaldehyde, propionaldehyde, butyraldehyde, valeraldehyde, caproaldehyde, heptaldehyde, stearaldehyde. Aromatic aldehydes which may be used include benzaldehyde and salicylaldehyde. Illustrative heterocyclic aldehydes for use herein are furfural and thiophene aldehyde, etc. Also useful are formaldehyde-producing reagents such as paraformaldehyde, or aqueous formaldehyde solutions such as formalin. Most preferred is formaldehyde or formalin.

[0032] Components i), ii), iii) and iv) are reacted under suitable Mannich reaction conditions to form a Mannich condensation reaction product.

[0033] The condensation reaction among the alkyl-substituted hydroxyaromatic compound, the amines and the aldehyde may be conducted at a temperature in the range of about 40° to about 200° C. The reaction can be conducted in bulk (no diluent or solvent) or in a solvent or diluent. Water is evolved and can be removed by azeotropic distillation during the course of the reaction. Typically, the Mannich additives are formed by reacting the alkyl-substituted hydroxyaromatic compound, the compound containing at least one primary or secondary amine group and at least one hydroxyl group, polyamine and aldehyde in the molar ratio of 1.0:0.2-1.8:0.2-1.8:1.0-4.0, respectively, preferably 1:0.5-1.5:0.5-1.5:1.5-2.5. The aldehyde is typically present in a molar amount at least equal to the total molar amount of the amine compounds present.

[0034] Although the novel reaction products of the present invention may be formed by reacting a mixture of amines (i) and (ii) with the hydroxyaromatic compound and an aldehyde, it is preferred, although not required, to 1) react the hydroxyaromatic compound with either amine (i) or amine (ii) and an aldehyde, and 2) react the reaction product of 1) with the other amine, either (ii) or (i) whichever was not used in step 1, and an aldehyde.

[0035] Thus one embodiment of the present invention comprises a process for preparing Mannich reaction products comprising:

- 1) reacting an alkyl-substituted hydroxyaromatic compound (iii), an aldehyde (iv) and a compound containing at least one primary or secondary amine group and at least one hydroxyl group (i); and
- 2) reacting the product formed in 1) with an aldehyde (iv) and a polyamine containing at least two amine groups wherein at least one of the amine groups is a primary or secondary amine (ii), and wherein component ii) is different from component i). In step 1) of the above process components iii), iv) and i) are preferably present in the

molar ratio of 1.0:0.2-3.6:0.2-1.8, respectively. In step 2), components iv) and ii) are preferably present in an amount of 0.2-3.6 moles of iv) and 0.2-1.8 moles of ii) per mole of alkyl-substituted hydroxyaromatic compound used in step 1).

[0036] Another embodiment of the present invention comprises a process for preparing Mannich reaction products comprising:

- 1) reacting an alkyl-substituted hydroxyaromatic compound (iii), an aldehyde (iv) and a polyamine containing at least two amine groups wherein at least one of the amine groups is a primary or secondary amine (ii); and
- 2) reacting the product formed in 1) with an aldehyde (iv) and a compound containing at least one primary or secondary amine group and at least one hydroxyl group (i), and wherein component i) is different from component ii).

[0037] In step 1) of the above process components iii), iv) and ii) are preferably present in the molar ratio of 1.0:0.2-3.6:0.2-1.8, respectively. In step 2), components iv) and i) are preferably present in an amount of 0.2-3.6 moles of iv) and 0.2-1.8 moles of i) per mole of alkyl-substituted hydroxyaromatic compound used in step 1).

[0038] When formulating the fuel compositions of this invention, the dispersant additives (with or without other additives) are employed in an amount effective to improve the detergency of the fuel. Generally speaking the fuels of this invention will contain, on an active ingredient basis, an amount of dispersant additive in the range of about 1 to about 1000 parts by weight of additive per million parts by weight of fuel.

[0039] An advantage of the present invention is that the additive reaction products do not detrimentally impact the combustion properties of the fuel. Further, the reaction products of the present invention can contribute lubricity benefits to the formulated fuel compositions. The improvements in the fuels lubricity will allow formulation with less, or even no, additional lubricity additive. Further, the improved fuel lubricity can reduce fuel pump wear.

[0040] The fuel compositions of the present invention may contain supplemental additives in addition to the reaction products described above. Said supplemental additives include supplemental dispersant/detergents, octane improvers, cetane improvers, antioxidants, carrier fluids, metal deactivators, dyes, markers, corrosion inhibitors, biocides,

antistatic additives, drag reducing agents, demulsifiers, dehazers, anti-icing additives, antiknock additives, anti-valve-seat recession additives, lubricity additives and combustion improvers.

[0041] Cyclopentadienyl manganese tricarbonyl compounds such as methylcyclopentadienyl manganese tricarbonyl are preferred combustion improvers because of their outstanding ability to reduce tailpipe emissions such as NOx and smog forming precursors and to significantly improve the octane quality of gasolines, both of the conventional variety and of the "reformulated" types.

[0042] The base fuels used in formulating the fuel compositions of the present invention include any base fuels suitable for use in the operation of spark-ignition or compression-ignition internal combustion engines such as diesel fuel, jet fuel, kerosene, leaded or unleaded motor and aviation gasolines, and so-called reformulated gasolines which typically contain both hydrocarbons of the gasoline boiling range and fuel-soluble oxygenated blending agents, such as alcohols, ethers and other suitable oxygen-containing organic compounds. Oxygenates suitable for use in the present invention include methanol, ethanol, isopropanol, t-butanol, mixed C1 to C5 alcohols, methyl tertiary butyl ether, tertiary amyl methyl ether, ethyl tertiary butyl ether and mixed ethers. Oxygenates, when used, will normally be present in the base fuel in an amount below about 25% by volume, and preferably in an amount that provides an oxygen content in the overall fuel in the range of about 0.5 to about 5 percent by volume.

[0043] Any middle-distillate fuel may be used in the present invention, however, high-sulfur content fuels typically do not require additional lubricity additives. In a preferred embodiment, the middle-distillate fuel is a diesel fuel having a sulfur content of up to about 0.2% by weight, more preferably up to about 0.05% by weight, as determined by the test method specified in ASTM D 2622-98.

[0044] The additives used in formulating the preferred fuels of the present invention can be blended into the base fuel individually or in various sub-combinations. However, it is preferable to blend all of the components concurrently using an additive concentrate (i.e., additives plus a diluent, such as a hydrocarbon solvent). The use of an additive concentrate takes advantage of the mutual compatibility afforded by the combination of ingredients when in the form of an additive concentrate. Also, the use of a concentrate reduces blending time and lessens the possibility of blending errors.

[0045] The examples given below illustrate the novel fuel compositions of the present invention. Unless otherwise specified, all proportions are given by weight. The following examples are not intended or should not be construed as limitations of the invention as presently claimed.

EXAMPLE I

[0046] The dispersants were prepared as follows: Dispersants A, B and C were prepared by reacting components (i) and (ii), set forth below, with a polyisobutenyl-substituted phenol and formaldehyde. The molar proportions of amine (i):phenol:formaldehyde in dispersant A were 1:1:1. The molar proportions of amine (i):amine (ii):phenol:formaldehyde in dispersants B and C were 1:1:1:2. Dispersants B and C were prepared by 1) reacting amine i), formaldehyde and the phenol at a molar ratio of 1:1:1, and 2) reacting the product of 1) with amine ii) and formaldehyde at a molar ratio of 1:1:1. The reactions were conducted in xylene at 145°C.

[0047] Dispersant A: Component i) was diethanolamine; and no component ii) was used.

[0048] Dispersant B: Component i) was diethanolamine; and component ii) was N,N-dimethyl-1,3-propanediamine.

[0049] Dispersant C: Component i) was diethanolamine; and component ii) was N-methyl piperazine.

[0050] The efficacy of the detergent/lubricity additives of the present invention was assessed using the Scuffing Load BOCLE (ball-on-cylinder lubricity evaluator) test (ASTM D 6078-97). The Scuffing Load BOCLE test allows discrimination and ranking of fuels of differing lubricity. The Scuffing test simulates the severe modes of wear failure encountered in fuel pumps and therefore provides results which are representative of how the fuel would behave in service. The load at which wear failure occurs is referred to as the scuffing load and is a measure of the inherent lubricity of the fuel. The scuffing load is primarily identified by the size and appearance of the wear scar on the ball, which is considerably different in appearance to that found under milder non-scuffing conditions. Fuels giving a high scuffing load on failure have better lubricating properties than fuels giving a low scuffing load on failure. All Scuffing Load BOCLE tests were conducted in a Jet A fuel containing 115 ppm w/w of the dispersant.

[0051] The only diesel detergency test that has met any degree of acceptance in the U.S. is the Cummins L10 test. A low sulfur No. 2D diesel fuel was used for the L 10 testing. The fuel containing 130 ppm w/w of a candidate additive is run in a Cummins L10 engine for 125 hours. At the end of the test, the injectors are removed and evaluated for plunger appearance. A trained rater inspects the plunger visually and assigns demerits following a Coordinating Research Council (CRC) protocol. CRC ratings for the six injectors are then averaged to give the test result. Lower CRC ratings indicate improved dispersancy/detergency.

Table 1

Dispersant	SLBOCLE Load (g)	Average CRC Rating
Base fuel only*	1600	26.7 ¹
A*	3000	27.6
B	2600	10.3
C	2800	18.5

* Comparative Examples not within the scope of the present invention

¹ Average of 3 runs

[0052] It is clear, upon examination of the data in Table 1, that the fuel compositions containing the additives of the present invention exhibit both improved lubricity and improved detergency as compared to base fuel alone or Mannich dispersants containing only the amine of component i).

[0053] As natural lubricity of gasoline is on a noticeable decline since refining has become more severe in order to produce "low emissions gasolines", a fuel additive is required by the industry to ensure that fuel pumps will accomplish an acceptable working lifetime. Although diesel fuel pumps and injectors operate under more stringent conditions than gasoline fuel pumps (15,000 to 30,000 psi vs. 40-60 psi for gasoline engines), there is a trend in the automotive industry to increase fuel system pressures, such as in the case of DIG engines (1000 to 2000 psi), and thus the demands made upon gasoline fuel pumps will increase.

[0054] In certain regions of the United States such as California, "low emission", highly oxygenated gasoline blends will put further abrasive demands upon fuel pumps. Gasoline fuel pump failures will continue to increase and therefore the industry is presently searching for an additive that will overcome this problem. The automotive industry is thus in need of a lubricity agent for fuels as they become harsher as a result of increased refining to achieve lower emissions. Further, commercialization of gasoline direct injection technology equipped vehicles with gasoline fuel pumps operating at much higher injection pressures will require careful consideration of gasoline lubricity properties. Thus, the present invention addresses these needs in an efficient and economical manner.

[0055] It is to be understood that the reactants and components referred to by chemical name anywhere in the specification or claims hereof, whether referred to in the singular or plural, are identified as they exist prior to coming into contact with another substance referred to by chemical name or chemical type (e.g., base fuel, solvent, etc.). It matters not what chemical changes, transformations and/or reactions, if any, take place in the resulting mixture or solution or reaction medium as such changes, transformations and/or reactions are the natural result of bringing the specified reactants and/or components together under the conditions called for pursuant to this disclosure. Thus the reactants and components are identified as ingredients to be brought together either in performing a desired chemical reaction (such as formation of the lubricity additive reaction products) or in forming a desired composition (such as an additive concentrate or additized fuel blend). It will also be recognized that the additive components can be added or blended into or with the base fuels individually per se and/or as components used in forming preformed additive combinations and/or sub-combinations. Accordingly, even though the claims hereinafter may refer to substances, components and/or ingredients in the present tense ("comprises", "is", etc.), the reference is to the substance, components or ingredient as it existed at the time just before it was first blended or mixed with one or more other substances, components and/or ingredients in accordance with the present disclosure. The fact that the substance, components or ingredient may have lost its original identity through a chemical reaction or transformation during the course of such blending or mixing operations is thus wholly immaterial for an accurate understanding and appreciation of this disclosure and the claims thereof.

[0056] As used herein the term "fuel-soluble" means that the substance under discussion should be sufficiently soluble at 20° C in the base fuel selected for use to reach at least the minimum concentration required to enable the substance to serve its intended function. Preferably, the substance will have a substantially greater solubility in the base fuel than this. However, the substance need not dissolve in the base fuel in all proportions.

[0057] This invention is susceptible to considerable variation in its practice. Therefore the foregoing description is not intended to limit, and should not be construed as limiting, the invention to the particular exemplifications presented hereinabove. Rather, what is intended to be covered is as set forth in the ensuing claims and the equivalents thereof permitted as a matter of law.

[0058] Patentee does not intend to dedicate any disclosed embodiments to the public, and to the extent any disclosed modifications or alterations may not literally fall within the scope of the claims, they are considered to be part of the invention under the doctrine of equivalents.

[0059] The said compound (i) is typically an amino-alcohol or an alkoxyated amine. Preferred amino-alcohols include

compounds of formula R_2-NH-R_1-OH wherein R_1 is a divalent C_1-C_6 alkyl group optionally substituted by 1 or 2 further hydroxy groups and R_2 is a hydrogen atom or a C_1-C_6 alkyl group which optionally carries 1, 2 or 3 hydroxy groups. Preferred alkoxylated amines include compounds of formula $Y-NR^1-Z$ wherein Y represents $R'(-(NR^1)-R^1)_n-$ and Z represents $-(R''O)_m-H$ wherein each R' is the same or different and is hydrogen or C_1-C_6 alkyl, for example methyl, provided that at least one R' is hydrogen, each R'' is the same or different and is C_1-C_6 alkylene, for example ethylene and propylene, n is from 0 to 10, preferably from 0 to 5, and m is from 1 to 10, preferably from 1 to 5.

[0060] Examples of suitable amine reactants (ii) include compounds of formula HNR_4R_5 wherein R_4 and R_5 are the same or different and each represent $-R^1-X_n$ wherein R^1 is a C_1-C_6 alkyl or C_2-C_6 alkenyl group, and X is a tertiary amino group, for example $-NR_1R_2$ wherein R_1 and R_2 are the same or different and each represent C_1-C_6 alkyl, or R_4 and R_5 , together with the nitrogen atom to which they are attached, together form a saturated 5- to 10-membered heterocyclic ring, which ring contains a nitrogen atom optionally substituted by a C_1-C_6 alkyl group, for example a methyl group.

[0061] Further suitable amine reactants (ii) are alkylene polyamines of formula $R'R''-N-(A-NR^1)_nR''$ wherein each A is the same or different and is a C_1-C_{12} alkylene group, preferably a C_1-C_4 alkylene group, and each R' and R'' are the same or different and represent hydrogen or C_1-C_{12} alkyl, for example C_1-C_4 alkyl, provided that at least one R' or R'' is hydrogen, and n is from 1 to 10.

[0062] Further suitable amine reactants (ii) are N,N -(di- C_1-C_{12} alkyl)amino- α , ω - C_3-C_6 alkylendiamines and compounds of formula $(XR)_3-C-R^1-NH_2$ wherein each X is the same or different and is as defined above, each R is the same or different and is a C_1-C_6 alkyl group, and R^1 is a direct bond or is an alkylene group having from 1 to 5 carbon atoms.

[0063] The hydroxyaromatic moiety in the said alkyl-substituted hydroxyaromatic compounds is typically a C_6-C_{10} hydroxyaromatic moiety. It typically carries from 1 to 3, for example 1 or 2, hydroxy groups. In addition to the hydroxy substituent(s), the C_6-C_{10} hydroxyaromatic moiety typically contains up to 3, for example 0, 1 or 2, further substituents. Suitable such further substituents include C_1-C_6 alkyl, for example methyl and ethyl, C_2-C_6 alkenyl, for example ethenyl, C_6-C_{10} aryl, for example phenyl and $-(C_1-C_6 \text{ alkyl})-(C_6-C_{10} \text{ aryl})$, for example benzyl and phenylethyl.

[0064] The alkyl substituents present on the above hydroxyaromatic moiety are typically derived from a polyolefin having a number average molecular weight of from about 500 to about 3000 (preferably from about 500 to about 2100) as determined by GPC. Typically, the polyolefin has a polydispersity of from 1 to 4, typically from 1 to 2, as determined by GPC.

[0065] Typically, the said at least one aldehyde (iv) is a compound of formula $R-CHO$, wherein R is a hydrogen atom, an alkyl group having from 1 to 20 carbon atoms, preferably from 1 to 10 carbon atoms, an alkenyl group having from 2 to 20 carbon atoms, preferably from 2 to 10 carbon atoms, a C_6-C_{10} aryl group, for example phenyl, a C_5-C_{10} cycloalkyl or cycloalkenyl group or a 5- to 10-membered heterocyclic group, for example furfuryl and thiophenyl. The moiety R is optionally substituted by 1 or more, for example 1, 2 or 3, C_1-C_6 alkyl, C_2-C_6 alkenyl, C_6-C_{10} aryl, hydroxy or amino groups.

Claims

1. A product obtainable by reacting:

- (i) a compound containing at least one primary or secondary amine group and at least one hydroxyl group;
- (ii) a polyamine containing at least two amine groups, which is different from (i), wherein at least one of the amine groups is a primary or secondary amine;
- (iii) an alkyl-substituted hydroxyaromatic compound; and
- (iv) at least one aldehyde.

2. A product according to claim 1, wherein the molar ratio compound (i) : compound (ii) : compound (iii) : compound (iv) is 0.2-1.8 : 0.2-1.8 : 1.0 : 1.0-4.0.

3. A product according to claim 1 or 2, wherein compound (i) comprises diethanolamine.

4. A product according to any one of the preceding claims, wherein compound (ii) comprises an alkylene polyamine.

5. A product according to any one of claims 1 to 3, wherein compound (ii) comprises N,N -dimethyl-1,3-propanediamine.

6. A product according to any one of claims 1 to 3, wherein compound (ii) comprises N -methyl piperazine.

7. A product according to any one of the preceding claims, wherein compound (iii) comprises an alkyl-substituted phenol.
8. A product according to claim 7, wherein said alkyl-substituted phenol is polybutylphenol.
9. A fuel composition comprising:
 - (a) a liquid hydrocarbon fuel; and, as a dispersant additive,
 - (b) a product according to any one of the preceding claims.
10. A fuel according to claim 9, wherein said dispersant (b) is present in an amount of from about 1 to about 1000 parts by weight per million parts by weight of the fuel.
11. A fuel according to claim 10, wherein said dispersant (b) is present in an amount of from about 10 to about 500 parts by weight per million parts by weight of the fuel.
12. A fuel according to claim 11, wherein said dispersant (b) is present in an amount of from about 30 to about 300 parts by weight per million parts by weight of the fuel.
13. A fuel according to any one of claims 9 to 12, wherein said liquid hydrocarbon fuel comprises a compression-ignition fuel selected from diesel, biodiesel, jet fuel and kerosene.
14. A fuel composition according to claim 13, wherein said compression-ignition fuel has a sulfur content below 0.2% by weight.
15. A fuel according to any one of claims 9 to 12, wherein said liquid hydrocarbon fuel comprises a spark-ignition fuel selected from gasoline and reformulated gasolines.
16. A fuel according to claim 15, which further comprises a polyisobutylene amine, a polyether amine and/or an oxygenate.
17. A fuel according to any one of claims 9 to 16, further comprising at least one further additive selected from supplemental dispersants/detergents, octane improvers, cetane improvers, carrier fluids, demulsifiers, antioxidants, antifoam agents, anti-icing additives, biocides, combustion improvers, alkali or alkaline-earth metal detergents, drag reducers, metal deactivators, lubricity additives, dyes, markers, odor masks, odorants and stability improvers.
18. A method of reducing deposits in a compression-ignition or spark-ignition combustion engine, which method comprises operating said engine on a fuel according to any one of claims 9 to 17.
19. Use of a product according to any one of claims 1 to 8, as a detergent in a hydrocarbon fuel.
20. Use of a product according to any one of claims 1 to 8, to improve the lubricity of a hydrocarbon fuel.
21. Use of a fuel according to any one of claims 9 to 17, in an engine designed so as to permit contact of the fuel with engine oil.
22. Use according to claim 21, wherein the engine contains an in-line diesel injection pump.
23. Use of a fuel according to any one of claims 9 to 12 and 15 to 17, in a direct injection gasoline engine and/or a gasoline engine having a submerged fuel pump in a fuel storage tank.
24. A process for preparing a product according to any one of claims 1 to 8, which process comprises:
 - 1) reacting an alkyl-substituted hydroxyaromatic compound (iii) as defined in any one of claims 1, 7 and 8, an aldehyde (iv) and a compound (i) containing at least one primary or secondary amine group and at least one hydroxyl group, as defined in any one of claims 1 and 3; and
 - 2) reacting the thus obtained product with an aldehyde (iv) and a polyamine (ii) containing at least two amine groups wherein at least one of the amine groups is a primary or secondary amine, as defined in any one of

claims 1 and 4 to 6, compound (ii) being different from compound (i).

25. A process according to claim 24, wherein in step (1) the molar ratio compound (i) : compound (iii) : compound (iv) is 0.2-1.8 : 1.0 : 0.2-3.6 and/or in step (2) compounds (iv) and (ii) are present in an amount of 0.2-3.6 moles of (iv) and 0.2-1.8 moles of (ii) per mole of alkyl substituted hydroxyaromatic compound used in step (1).

26. A process for preparing a product according to any one of claims 1 to 8, which process comprises:

1) reacting an alkyl-substituted hydroxyaromatic compound (iii) as defined in any one of claims 1, 7 and 8, an aldehyde (iv) and a polyamine (ii) containing at least two amine groups wherein at least one of the amine groups is a primary or secondary amine, as defined in any one of claims 1 and 4 to 6; and

2) reacting the thus obtained product with an aldehyde (iv) and a compound containing at least one primary or secondary amine group and at least one hydroxyl group (i), as defined in any one of claims 1 and 3, compound (i) being different from compound (ii).

27. A process according to claim 26, wherein in step (1) the molar ratio compound (ii) : compound (iii) : compound (iv) is 0.2-1.8 : 1.0 : 0.2-3.6 and/or in step (2) compounds (iv) and (i) are present in an amount of 0.2-3.6 moles of (iv) and 0.2-1.8 moles of (i) per mole of alkyl-substituted hydroxyaromatic compound (iii) used in step (1).



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EUROPEAN SEARCH REPORT

Application Number
EP 00 30 7444

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	EP 0 744 458 A (ETHYL CORP) 27 November 1996 (1996-11-27) * page 5, line 3 - line 21; claims 1,7,12,17 * * page 5, line 45 - line 51 * ---	1,2, 7-19, 24-27	C10L1/22 C08F8/32 C08G14/06 C10L10/00 C10L10/04 C10L1/14
X	US 4 487 852 A (BRENNAN MICHAEL E ET AL) 11 December 1984 (1984-12-11) * claims 1-9 * ---	1-3,7	
X	US 4 599 371 A (LOCH WERNER ET AL) 8 July 1986 (1986-07-08) * column 6, line 44 - line 45; claim 1 * ---	1,3	
A	US 4 231 759 A (UDELHOFEN JOHN H ET AL) 4 November 1980 (1980-11-04) * the whole document * ---	3,5,15, 17-19	
A	EP 0 807 676 A (ETHYL PETROLEUM ADDITIVES LTD) 19 November 1997 (1997-11-19) * page 7, line 15 - line 20 * ---	20	TECHNICAL FIELDS SEARCHED (Int.Cl.7)
A	US 5 634 951 A (COLUCCI WILLIAM J ET AL) 3 June 1997 (1997-06-03) * the whole document * ---	5,15, 17-19	C10L C08F C08G C10M
A	PATENT ABSTRACTS OF JAPAN vol. 1999, no. 02, 26 February 1999 (1999-02-26) & JP 10 298258 A (ASAHI ORGANIC CHEM IND CO LTD), 10 November 1998 (1998-11-10) * abstract * ---	1,24	
A	EP 0 182 940 A (MOBIL OIL CORP) 4 June 1986 (1986-06-04) * the whole document * --- -/-	1,3,20	
The present search report has been drawn up for all claims			
Place of search: THE HAGUE		Date of completion of the search: 12 December 2000	Examiner: De La Morinerie, B
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document</p>			

EPO FORM 1503 (3-92) (P4001)



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 00 30 7444

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
A	EP 0 410 577 A (PETROLITE CORP) 30 January 1991 (1991-01-30) * the whole document *	1,13,15	
A	GB 2 010 324 A (CHEVRON RES) 27 June 1979 (1979-06-27) * page 1, line 110 - line 125 *	1,13	
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 12 December 2000	Examiner De La Morinerie, B
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X: particularly relevant if taken alone Y: particularly relevant if combined with another document of the same category A: technological background O: non-written disclosure P: intermediate document</p> <p>T: theory or principle underlying the invention E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons &: member of the same patent family, corresponding document</p>			

EPO FORM 503 03 82 (P04C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 00 30 7444

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

12-12-2000

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
EP 0744458	A	27-11-1996	US	5512067 A	30-04-1996
US 4487852	A	11-12-1984	NONE		
US 4599371	A	08-07-1986	DE	3329693 A	07-03-1985
			AT	28181 T	15-07-1987
			DE	3464589 D	13-08-1987
			EP	0134019 A	13-03-1985
US 4231759	A	04-11-1980	NONE		
EP 0807676	A	19-11-1997	US	6086645 A	11-07-2000
			CA	2204806 A	17-11-1997
			JP	10053777 A	24-02-1998
			KR	228992 B	01-11-1999
US 5634951	A	03-06-1997	AU	2473097 A	11-12-1997
			EP	0811672 A	10-12-1997
			JP	10060460 A	03-03-1998
			SG	50843 A	20-07-1998
			US	5725612 A	10-03-1998
JP 10298258	A	10-11-1998	NONE		
EP 0182940	A	04-06-1986	AT	51861 T	15-04-1990
			AU	584111 B	18-05-1989
			AU	4629785 A	26-02-1987
			DE	3481911 D	17-05-1990
			ZA	8506322 A	25-03-1987
EP 0410577	A	30-01-1991	CA	2019320 A	27-01-1991
			US	5047069 A	10-09-1991
GB 2010324	A	27-06-1979	US	4166726 A	04-09-1979
			AU	521424 B	01-04-1982
			AU	4121578 A	08-05-1980
			BE	872543 A	30-03-1979
			BR	7808275 A	14-08-1979
			CA	1114609 A	22-12-1981
			DE	2853543 A	21-06-1979
			ES	476045 A	01-11-1979
			FR	2411882 A	13-07-1979
			IT	1104588 B	21-10-1985
			JP	1403916 C	09-10-1987
			JP	54087706 A	12-07-1979
			JP	62010279 B	05-03-1987

EPO FORM P4459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.

EP 00 30 7444

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

12-12-2000

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
GB 2010324 A		MX 4538 E	03-06-1982
		NL 7812133 A, B,	19-06-1979
		PH 13904 A	27-10-1980
		SE 438687 B	29-04-1985
		SE 7812741 A	17-06-1979
		ZA 7806072 A	31-10-1979
<hr/>			

EPO FORM P/4149

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82